



Pore Architecture of Zeolitic Imidazolate Frameworks: An Investigation using Positron Annihilation Spectroscopy

Sandeep Kumar Sharma

RCD, BARC

skumars@barc.gov.in

Introduction

- MOFs, ZIFs and their pore architecture
- Issues with pore analysis of ZIFs

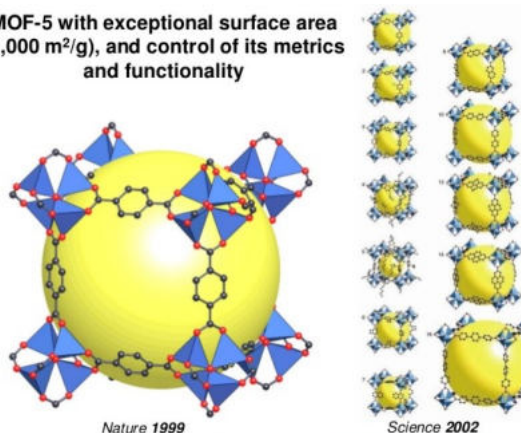
Application of positron annihilation spectroscopy in ZIFs

- Effect of external pressure
- Crystal size induced modifications
- Mixed ligand induced tuning of pore architecture
- Positronium diffusion in ZIF-8 films

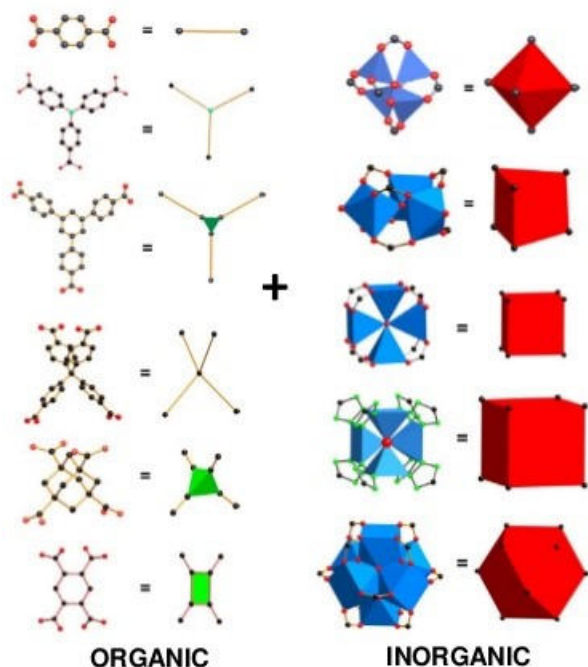
Conclusions and outlook

Metal Organic Frameworks

MOF-5 with exceptional surface area (3,000 m²/g), and control of its metrics and functionality



RETICULAR SYNTHESIS OF MOFs



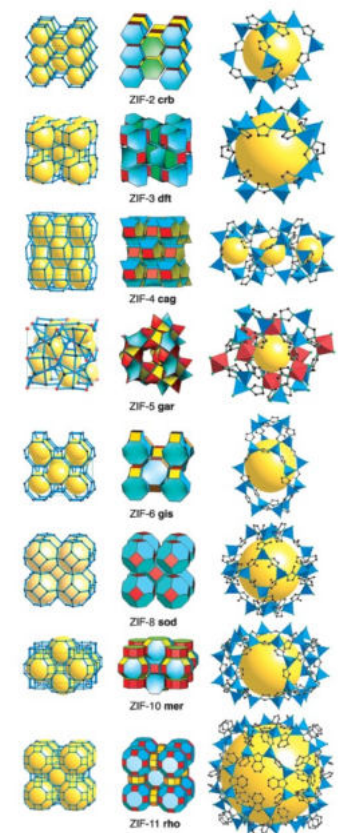
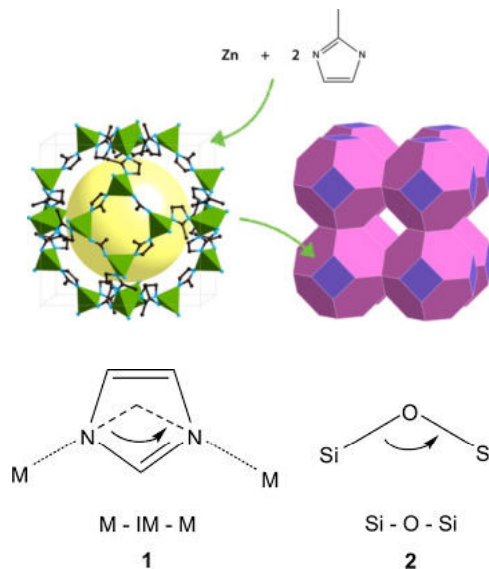
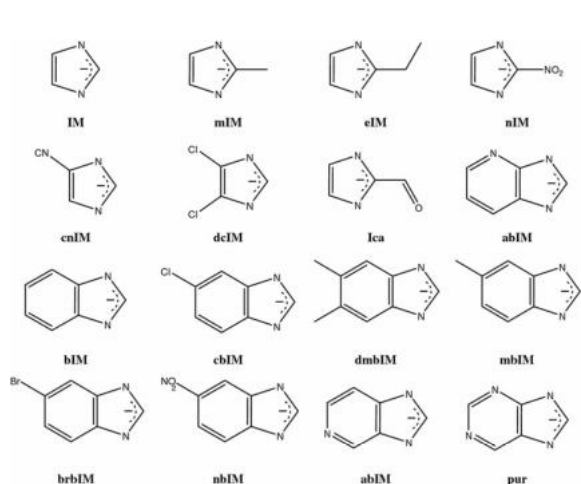
- MOFs are highly porous with very high surface area.
- Porosity is due to crystal structure.
- Pore size can be tuned using different metals and ligands.
- MOFs are thermally stable.

→ MOFs

- MOFs provide immense possibilities as a result of infinite combination of metals and ligands.

Zeolitic Imidazolate Framework

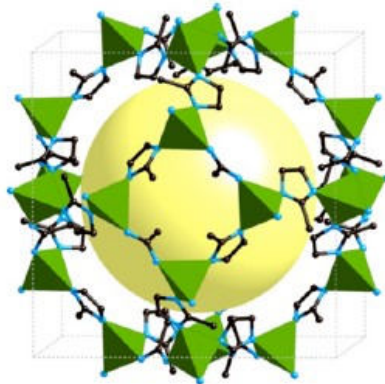
- Topology of Zeolites; M-Im-M bond similar to Si-O-Si bond



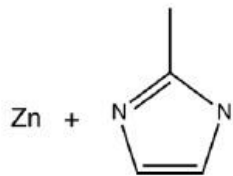
ZIF- <i>n</i>	Composition	Net*	Zeolite [†]	T/V_1^{\ddagger} nm ⁻³	d_1^{\S} Å	M^{\parallel}
ZIF-1	Zn(IM) ₂	crb	BCT	3.64	6.94	12
ZIF-2	Zn(IM) ₂	crb	BCT	2.80	6.00	12
ZIF-3	Zn(IM) ₂	dft	DFT	2.66	8.02	16
ZIF-4	Zn(IM) ₂	cag	—	3.68	2.04	20
ZIF-5	In ₂ Zn ₃ (IM) ₁₂	gar	—	3.80	3.03	20
ZIF-6	Zn(IM) ₂	gls	GIS	2.31	8.80	20
ZIF-7	Zn(PhIM) ₂	sod	SOD	2.50	4.31	24
ZIF-8	Zn(MeIM) ₂	sod	SOD	2.47	11.60	24
ZIF-9	Co(PhIM) ₂	sod	SOD	2.51	4.31	24
ZIF-10	Zn(IM) ₂	mer	MER	2.25	12.12	24
ZIF-11	Zn(PhIM) ₂	rho	RHO	2.01	14.64	48
ZIF-12	Co(PhIM) ₂	rho	RHO	2.01	14.64	48

ZIF-8

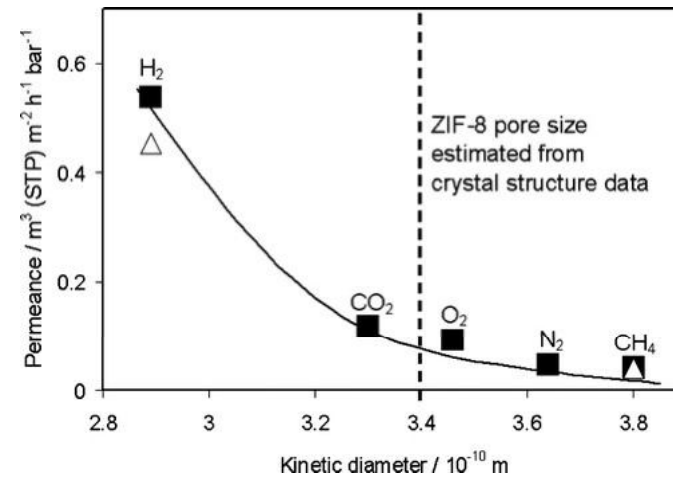
Single crystallography: Central cavity of 1.16 nm connected through 0.34 nm



ZIF-8 (SOD)



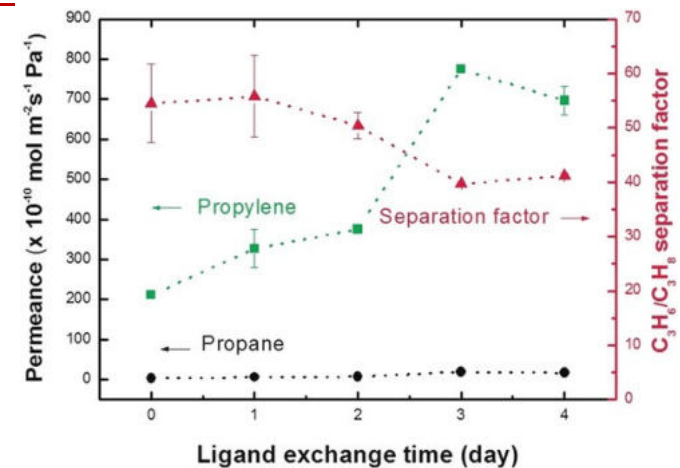
CO_2 (0.33) < Pore aperture < CH_4 (0.38)



J. Am. Chem. Soc. 2009, 131, 16000-16001

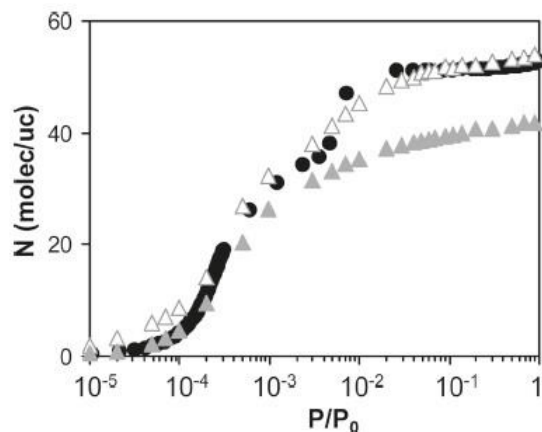
Poor selectivity

Large size molecules are passing through.

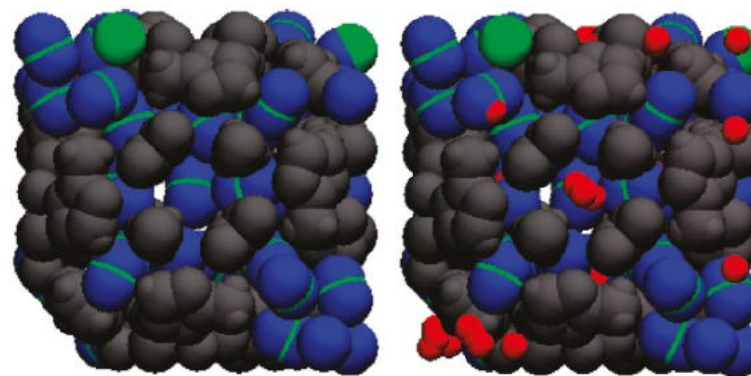


Angew. Chem., 2018, 57, 156-161

Gate Opening: Flexible framework

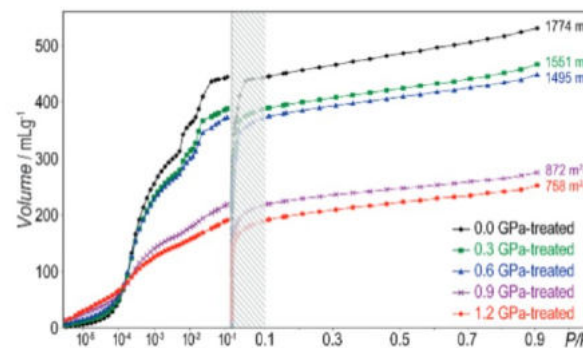
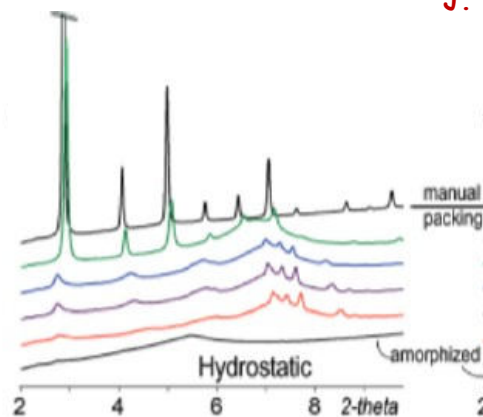
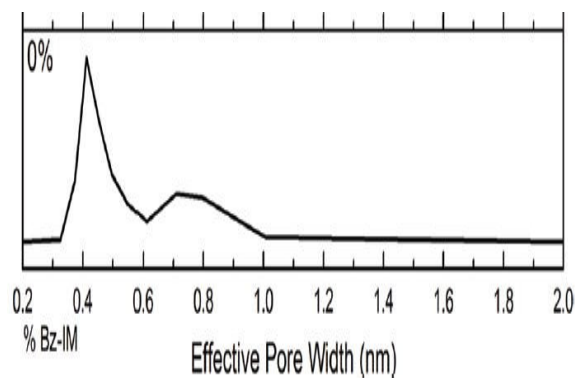


N₂ Adsorption at 77 K



Structural Changes at high pressure

J. Am. Chem. Soc. 2011, 133, 8900-8902



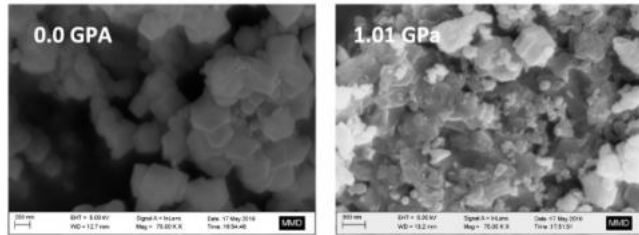
Pore size determined using N₂ adsorption does not match the crystallographic data.

Pressure induced Irreversible amorphization
Amorphous phase remains porous.

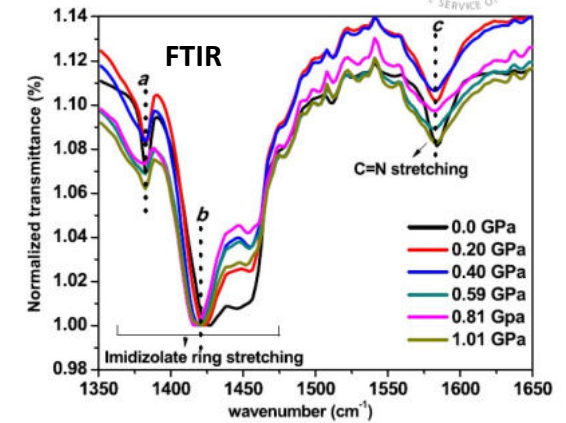
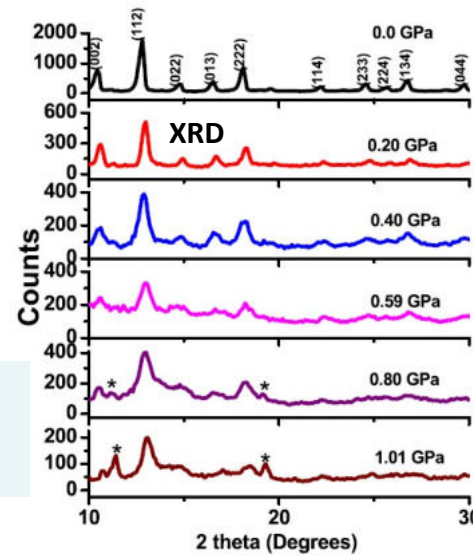
Chem. Mater. 2012, 24, 1930-1936

J. Am. Chem. Soc. 2009, 131, 17546-17547

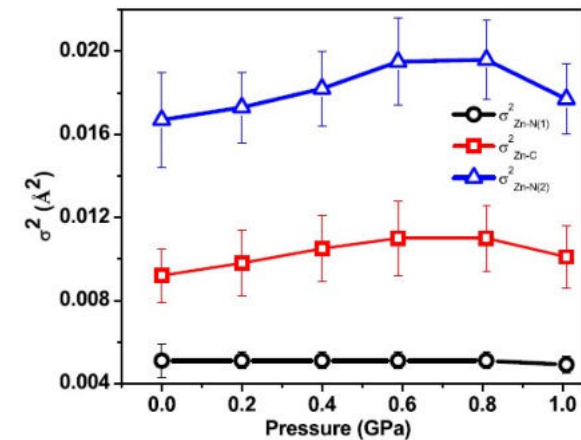
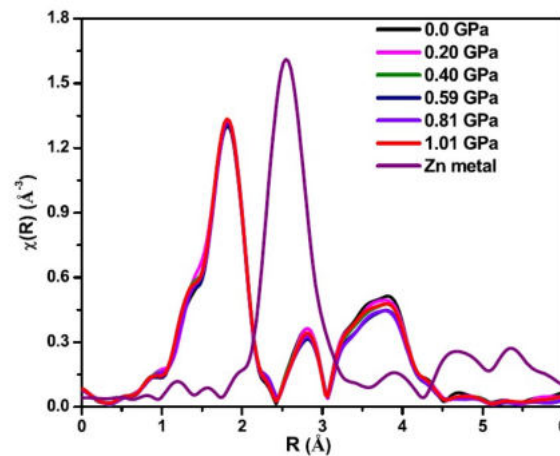
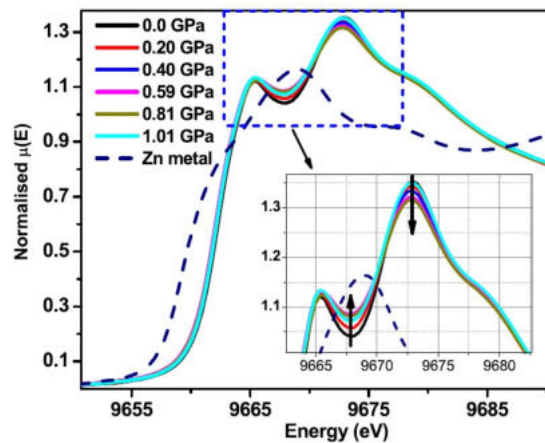
Compression induced change in porosity



- Irreversible amorphization
- Deformation in Imidazolate ring bonding

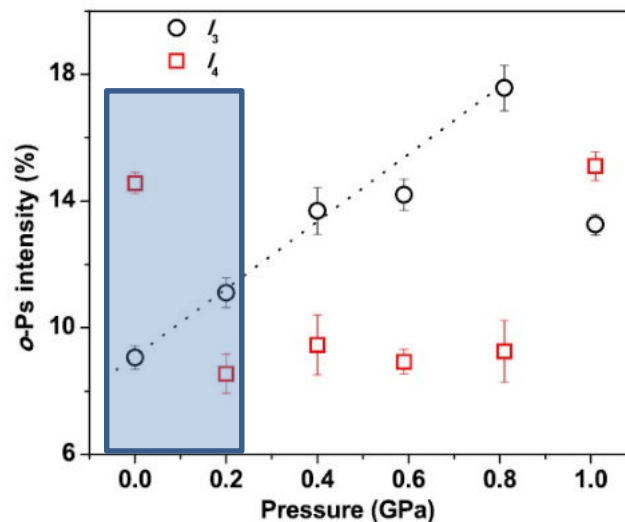
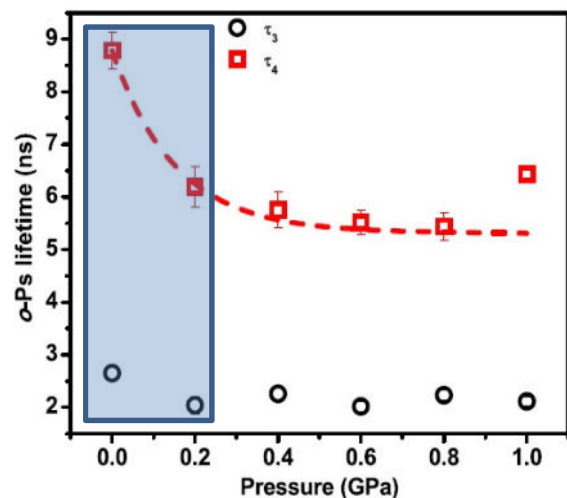


X-ray Absorption spectroscopy:



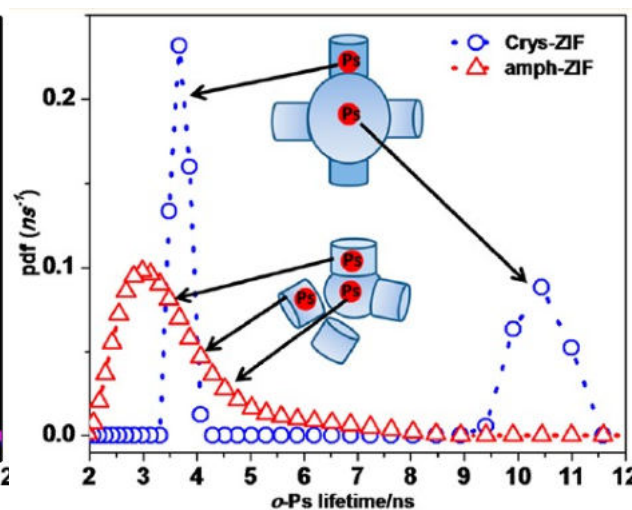
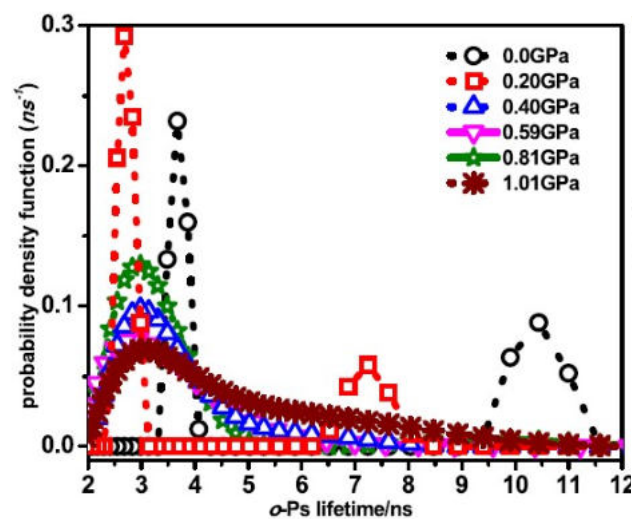
- Retention of Zn-N tetrahedral arrangement in amorphous state.

Positron annihilation lifetime measurements:



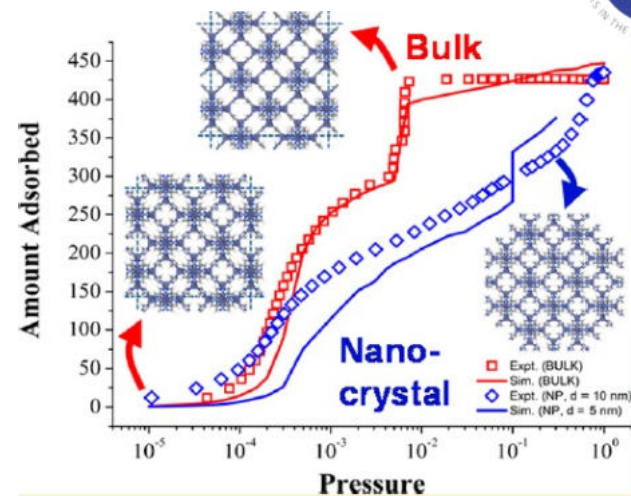
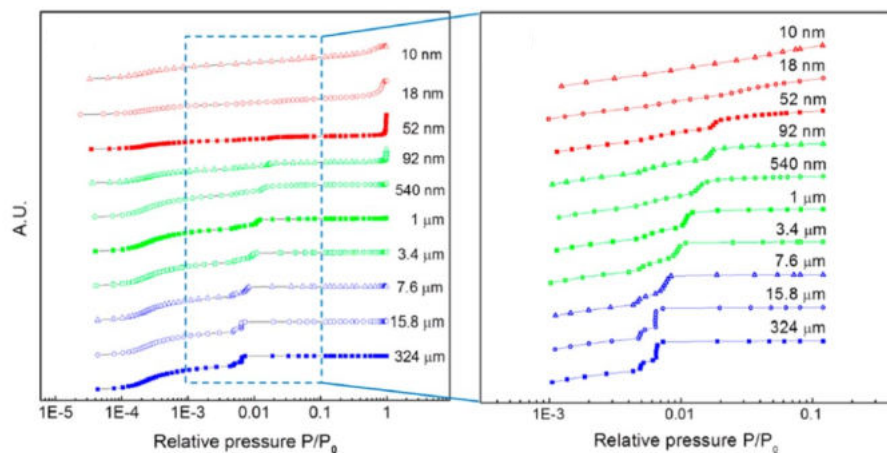
Lower pressure:
Partial collapse of open volume

Higher pressure:
Cataclysmic modifications



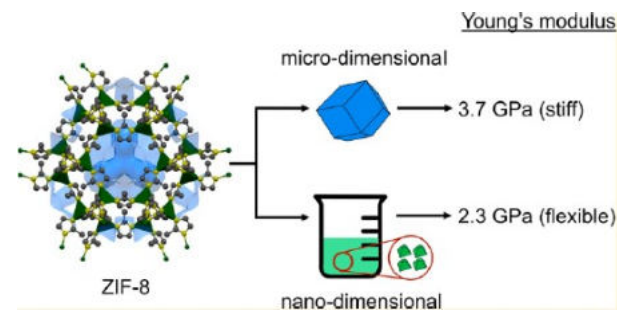
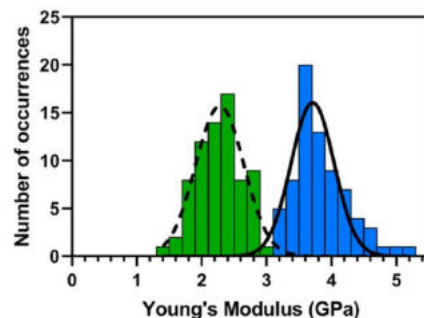
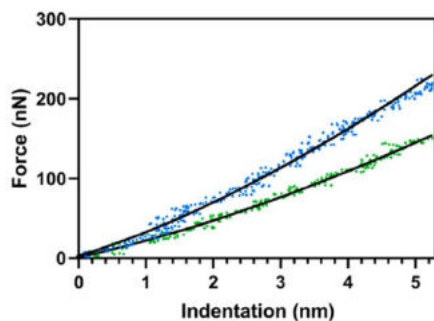
Amorphous phase:
Continuous random network of open volume of broader size distribution

Crystal size dependent porosity



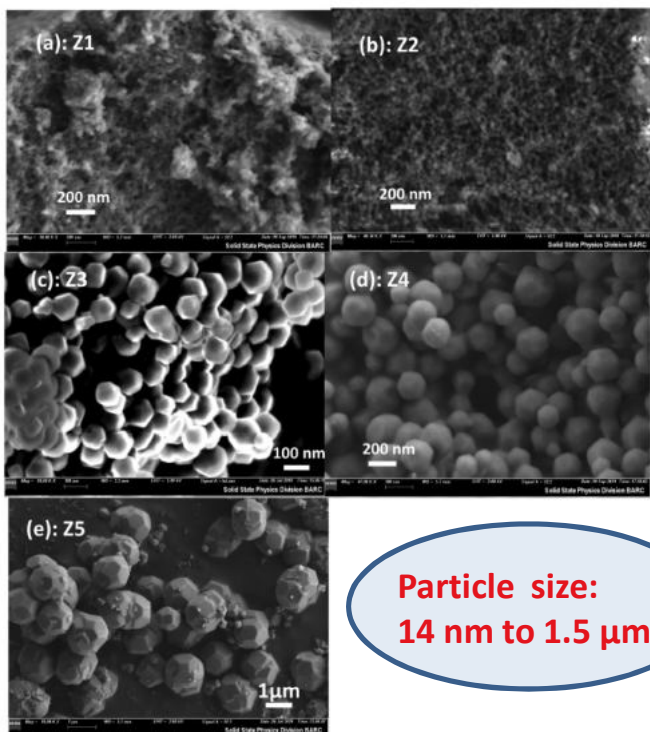
- Flexibility is reduced with reduction in crystal size
- Pore size distribution of solvated crystals ??

[J. Phys. Chem. C 118 \(2014\) 20727-20733](#)

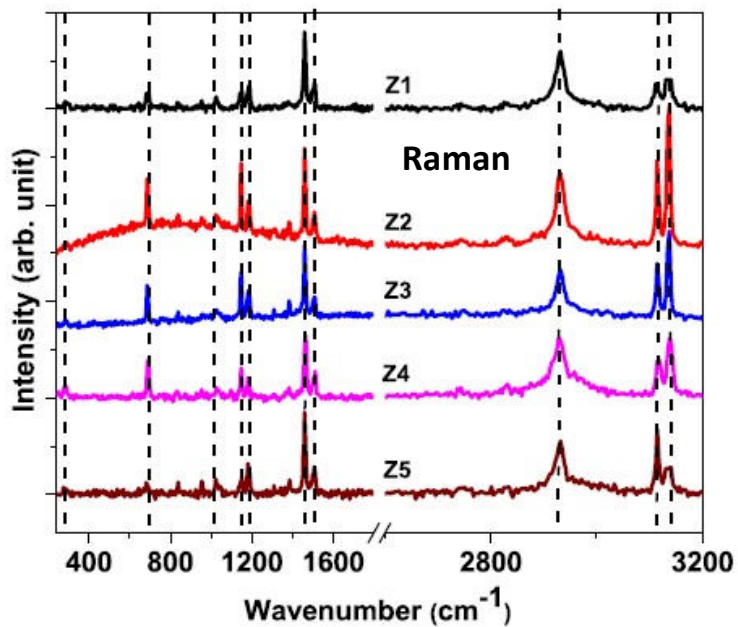
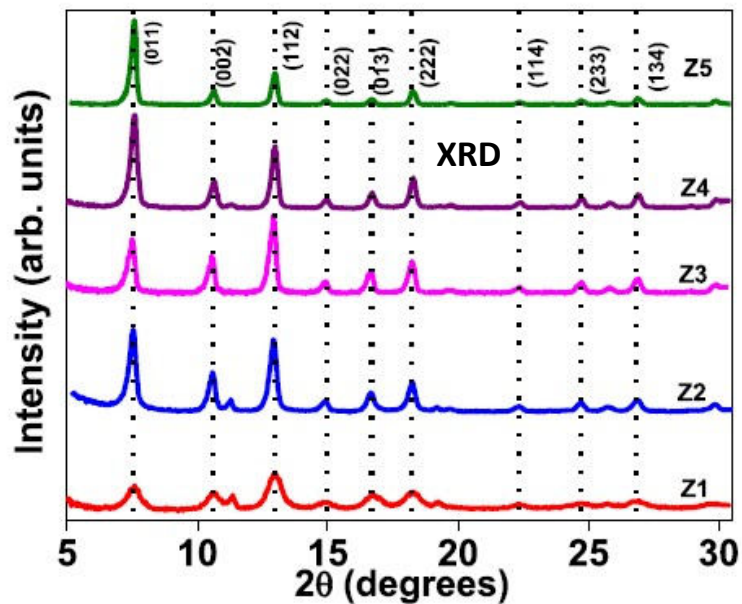


- Nanocrystals are more flexible.

[ACS Nano Letters 19 \(2019\) 6140-6143](#)

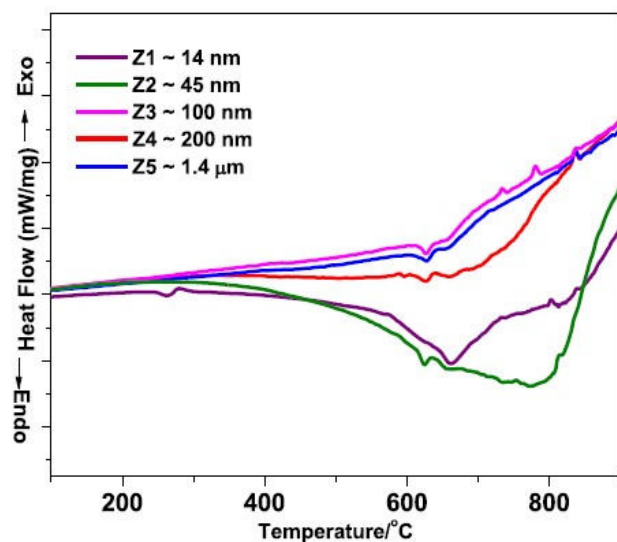
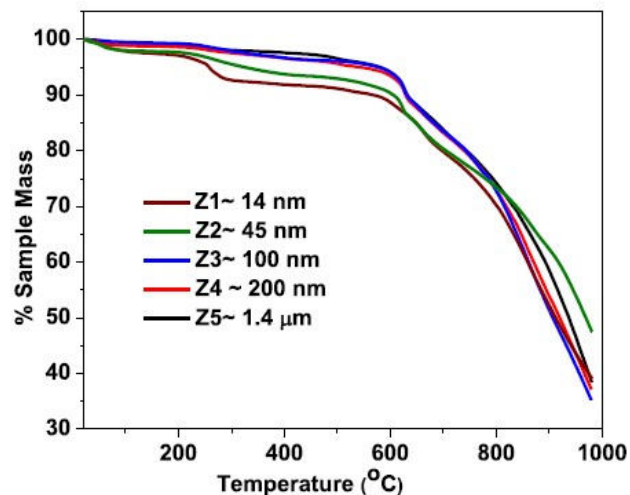


**Particle size:
14 nm to 1.5 μm**



- Phase pure ZIF-8 of 14 nm to 1.5 μm were synthesized.
- Chemical bonding and structure are intact.

Thermal analysis



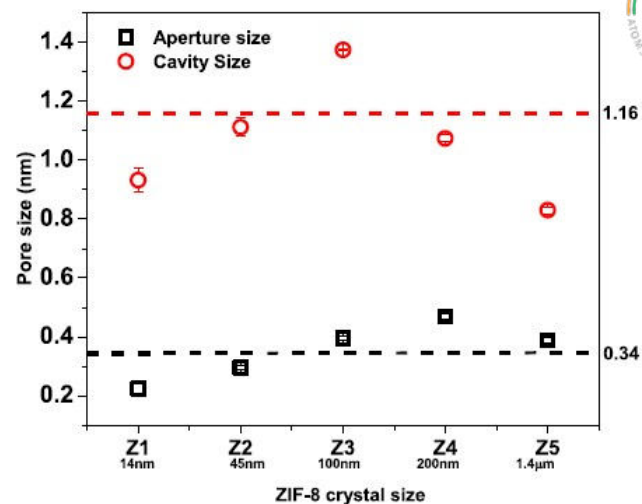
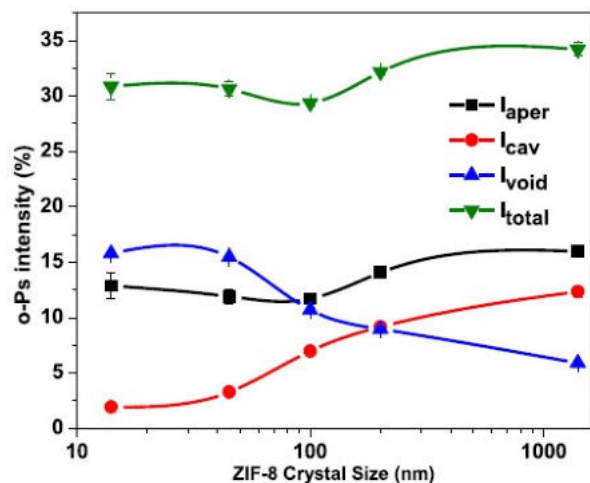
- Decomposition temperature remains nearly invariant.
- Thermal stability of nanocrystals is similar to larger crystals.

- No impurity phase.
- No additional phase transition.
- Decomposition of ZIF-8 ~ 750°C.

- Nanocrystals shows broader endothermic peak corresponding to decomposition.

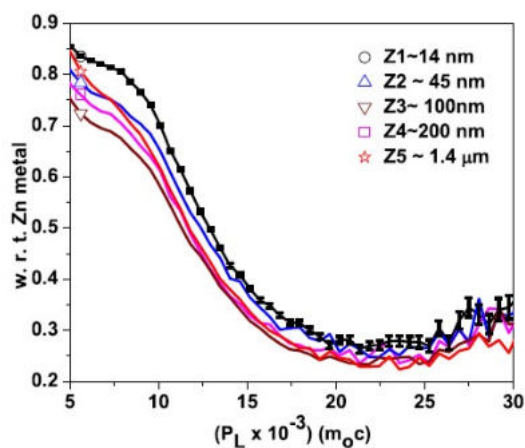
- Surface characteristics and pore architecture of nanocrystals are different.

Positron annihilation lifetime measurements:

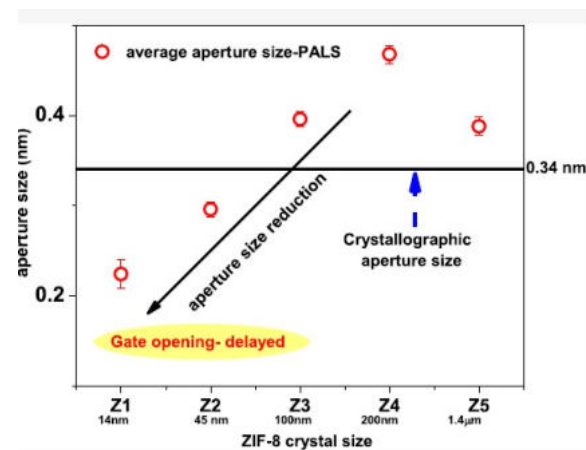


Pore interconnectivity is altered with crystal size.

- Average size is different from crystallographic size.
- Defective crystals during synthesis.

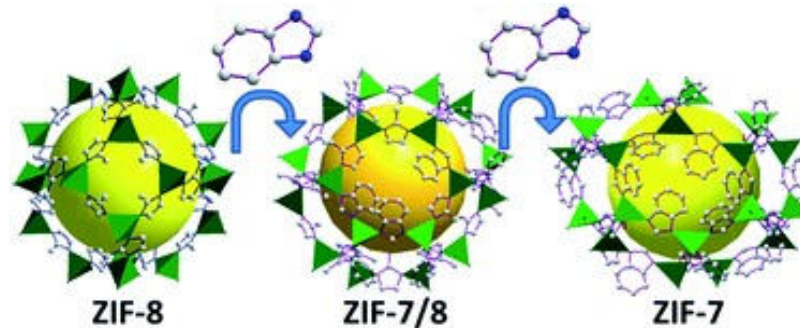


Nanocrystals have Zn enriched surface.

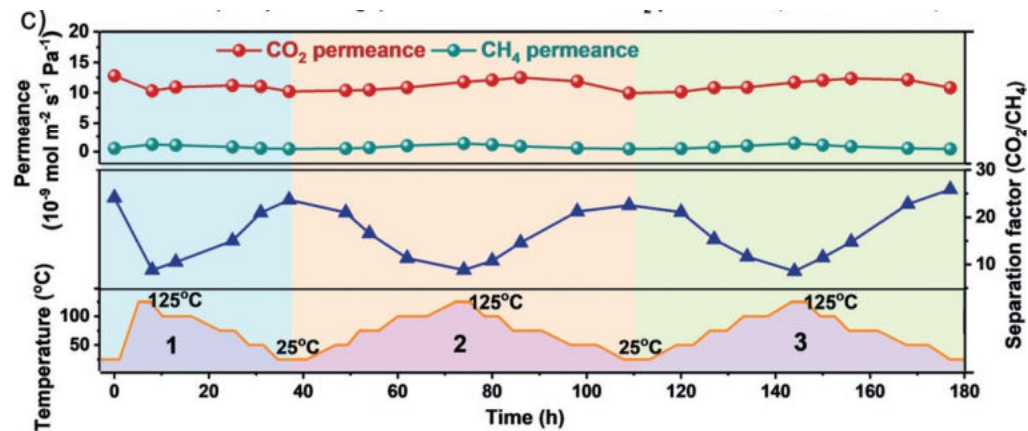


Delayed gate opening due to restricted pore aperture.

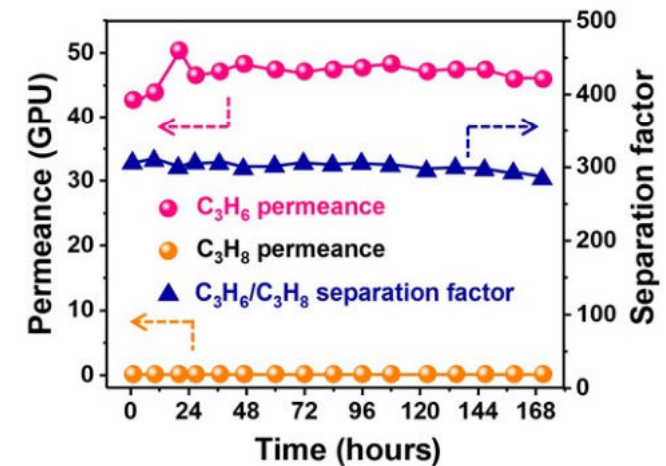
Pore tuning with mixed ligand strategy



[J. Mater. Chem. A 5 \(2017\) 25601](#)

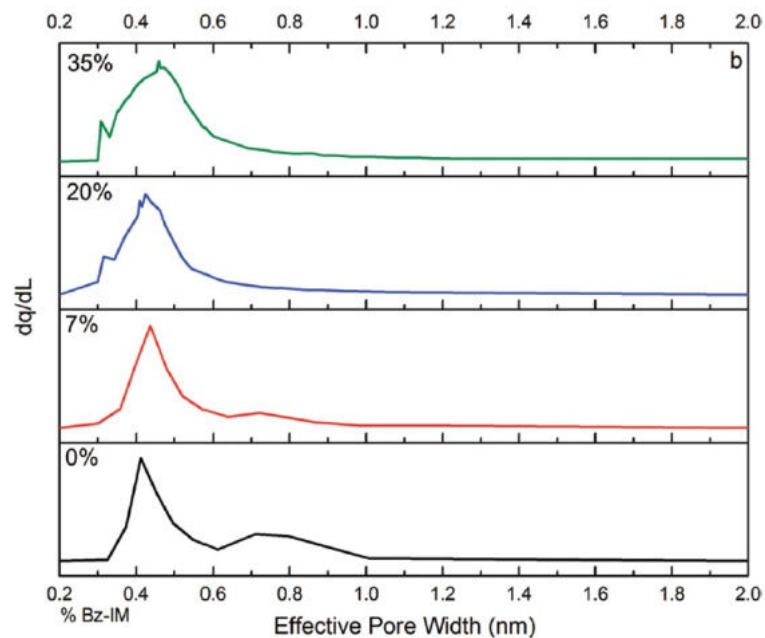


[Angew. Chem. 58 \(2019\) 327-331](#)

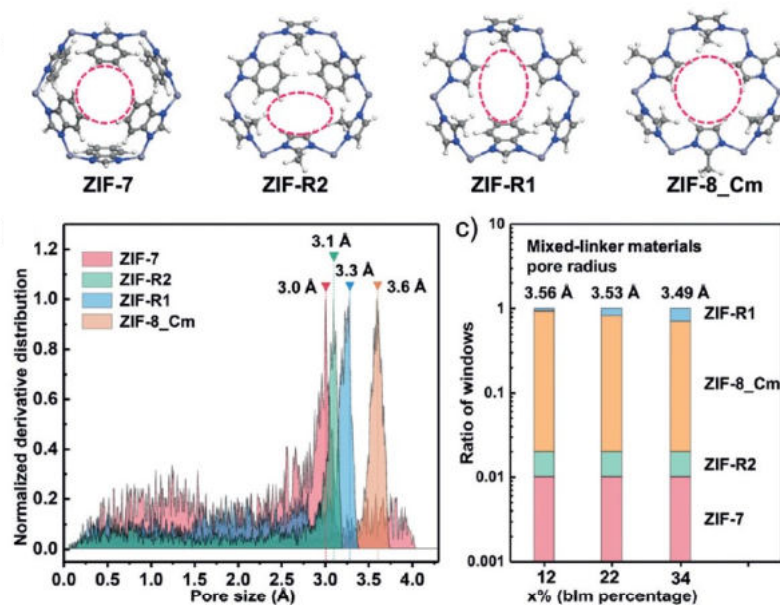


[Science Advances 4 \(2018\) eaau 1393](#)

Pore tuning of a stiffened phase using mixed ligand strategy is possible.



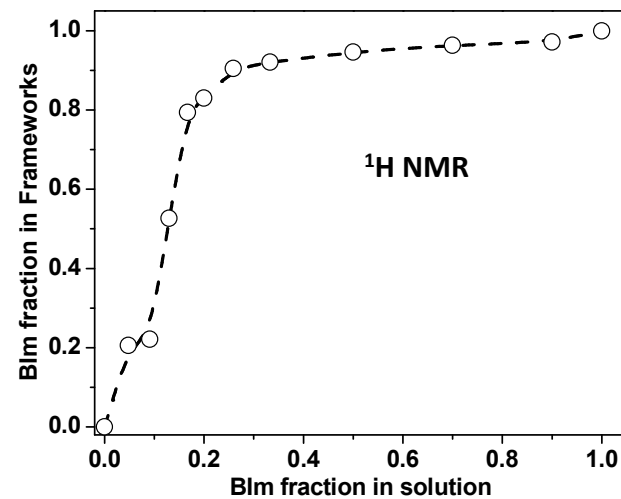
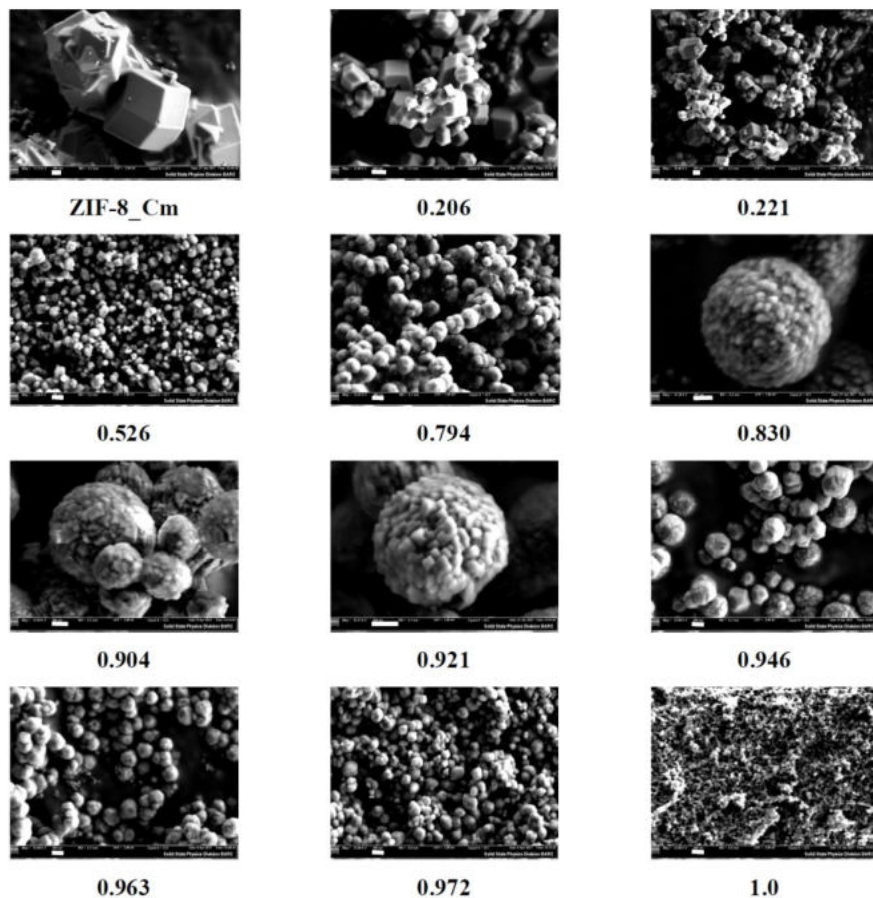
Chem. Mater. 2012, 24, 1930-136



Angew. Chem. 58 (2019) 327-331

- Pore size determined using gas adsorption is not absolute due to framework-gas interaction induced pore openings.
- Pore size from theoretical modeling does not represent the average pore sizes.

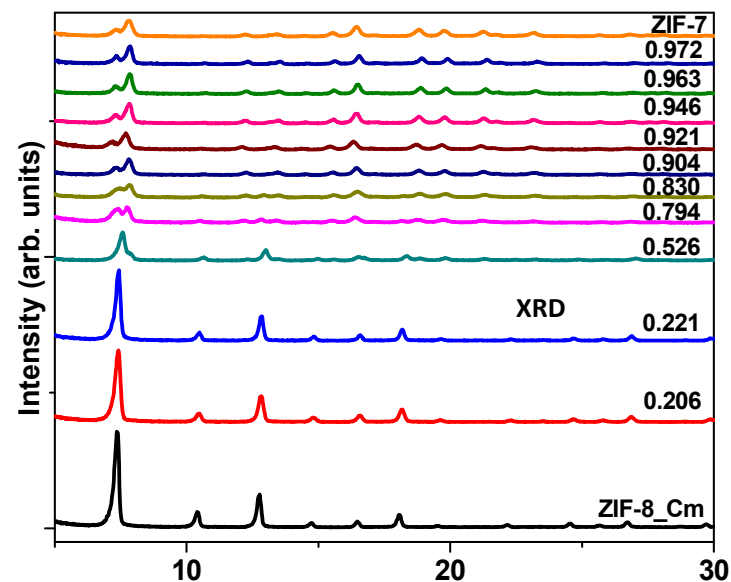
Morphology of ZIF-7_x-8 frameworks:

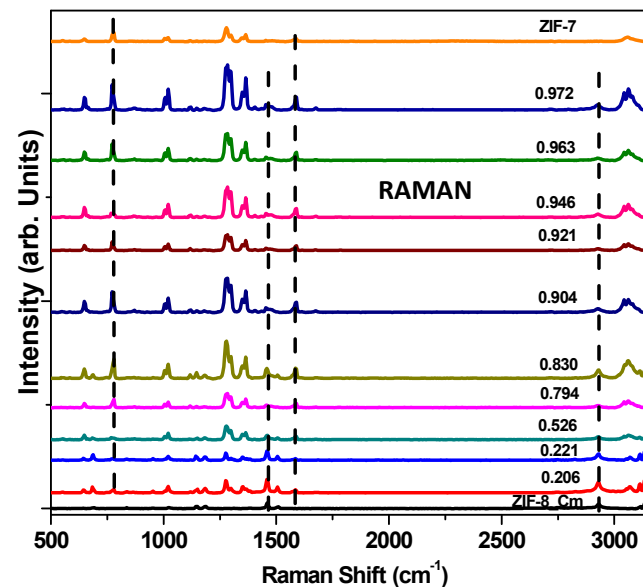
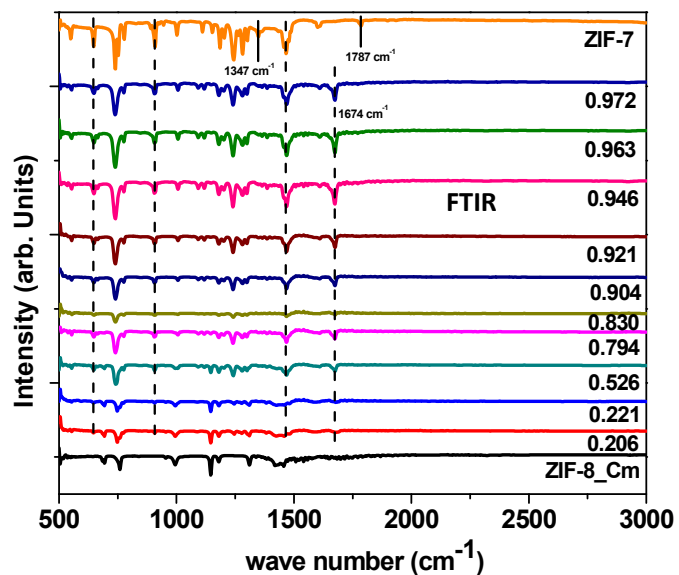


blm is incorporated in ZIFs preferentially.

With increase in blm fraction in frameworks

- Morphology changes drastically.
- Phase changes from ZIF-8_Cm to ZIF-7 (R3).

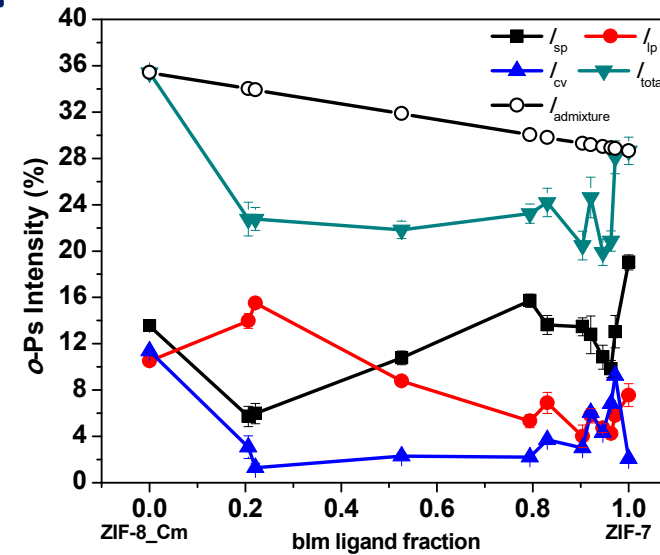
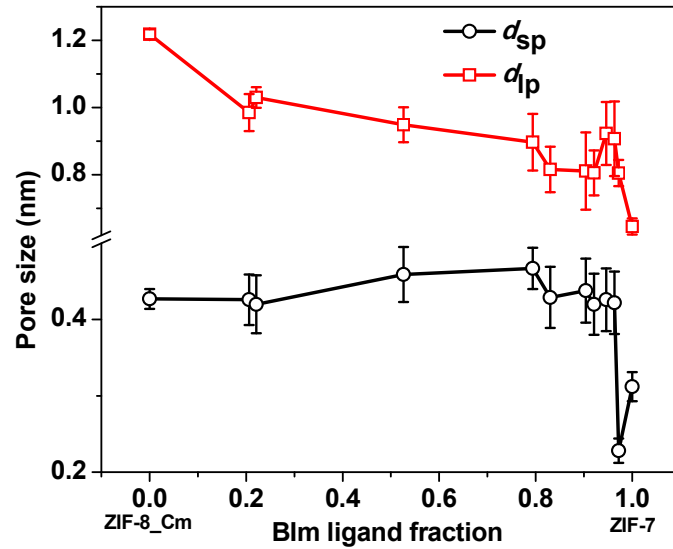




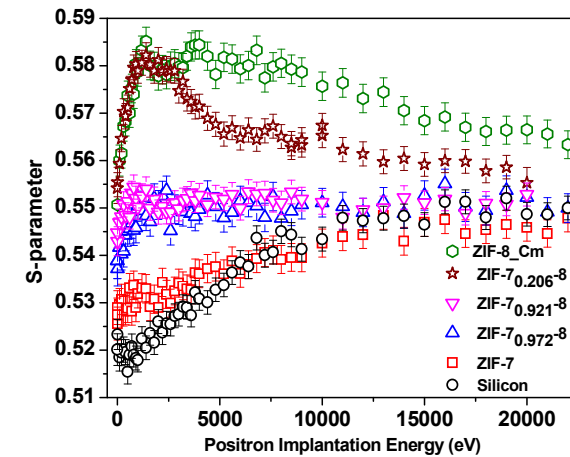
- Peaks from blm and 2-mlm are present in all the samples.
- NMR also confirms presence of both the ligands.
- Single morphology of each sample.

The synthesized ZIF-7_x-8 frameworks are single phase materials having randomly distributed ligand and not the mixture.

Positron annihilation measurements:

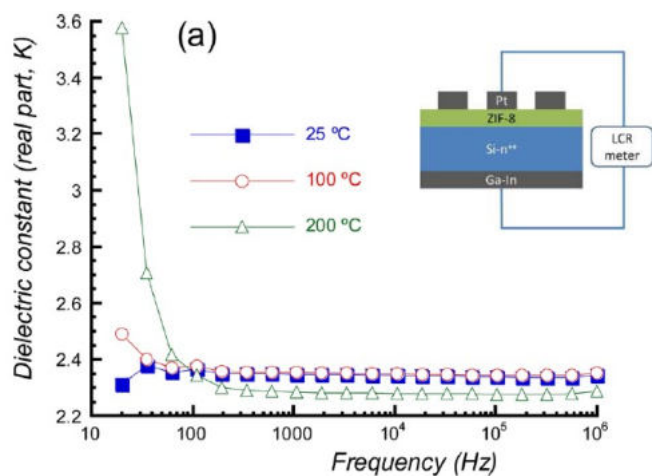
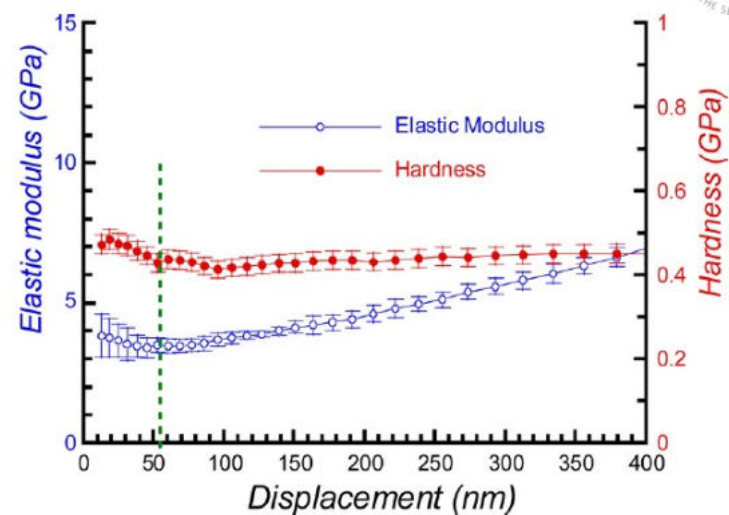
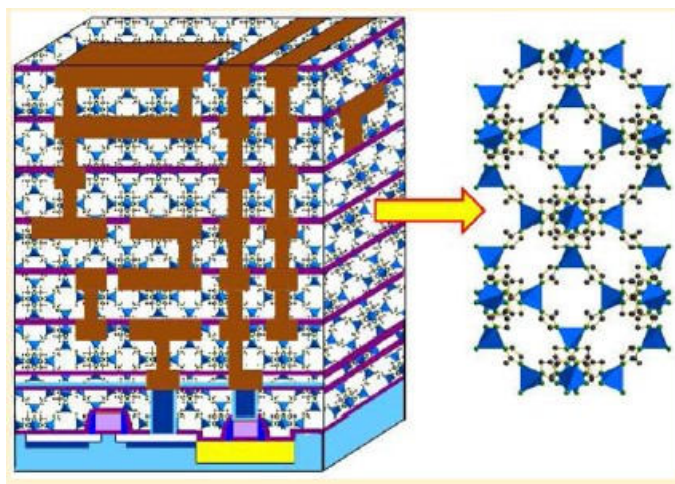


- The pore size for pure ZIF-7 and ZIF-8_Cm match very well.
- The pore size varies with blm ligand incorporation.
- Intensity does not follow the admixture rule of two components confirming that material is single phase.
- Pore interconnectivity is also hampered in mixed ligand network.



Sharma et al. *Mic. Meso. Mater.* (Revision submitted)

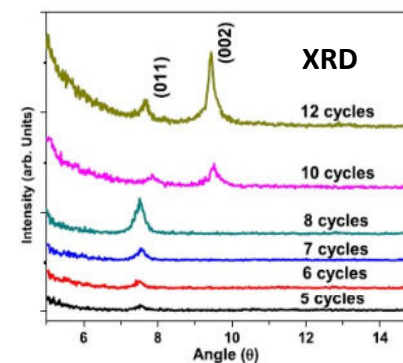
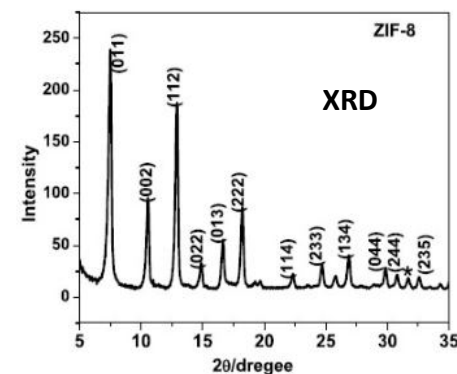
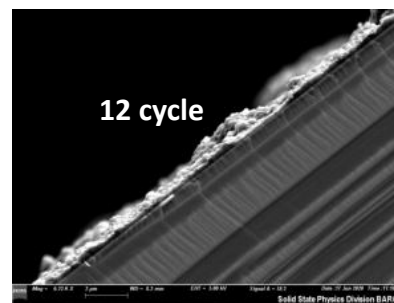
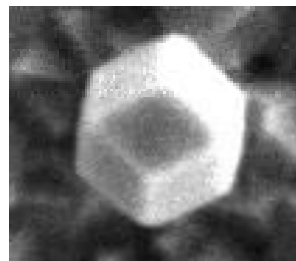
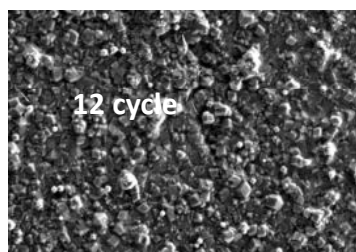
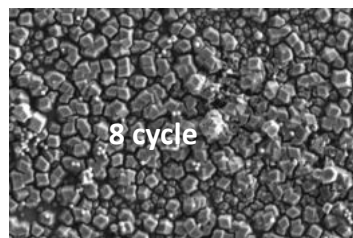
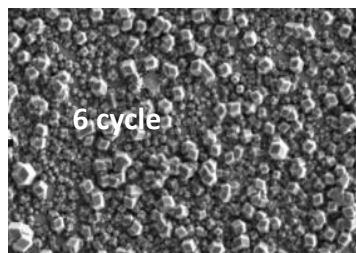
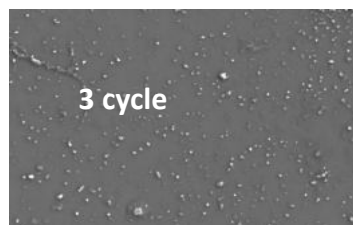
ZIF films as low k dielectric



Dielectric constant = 2.33 at 100 kHz

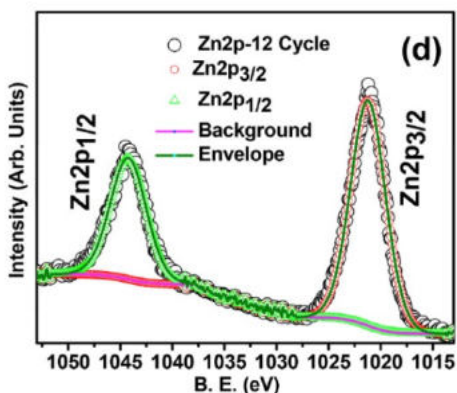
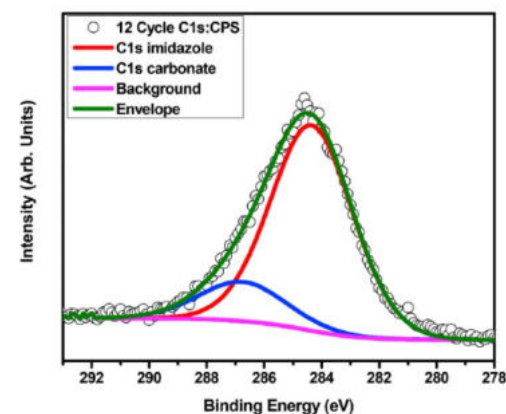
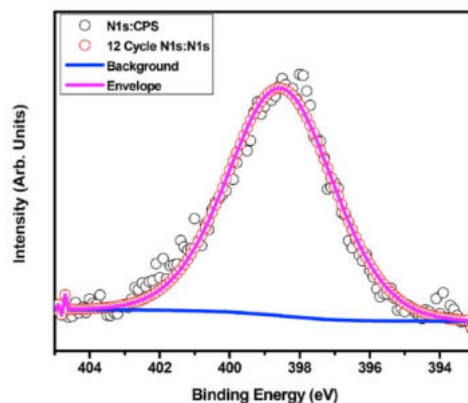
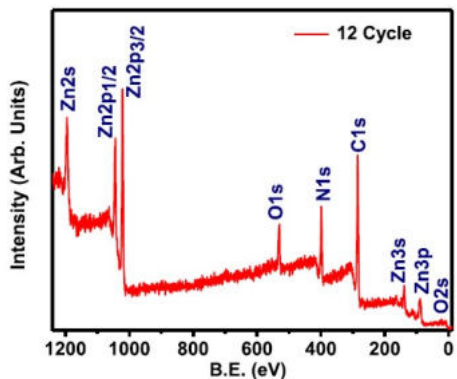
Elastic modulus > 3 GPa

Dielectric constant and applicability of the films will depend on the pore architecture.



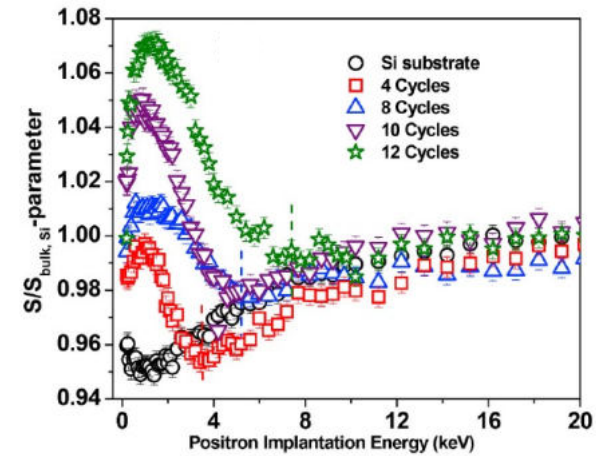
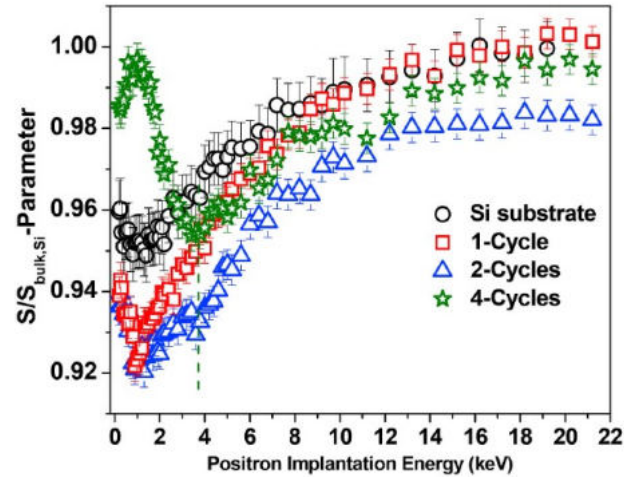
- ZIF-8 can be deposited at room temperature.
- ZIF-8 nuclei density increases in initial cycles and thickness growth occurs in later cycles.
- Preferential growth occurs in (002) direction.

XPS analysis:

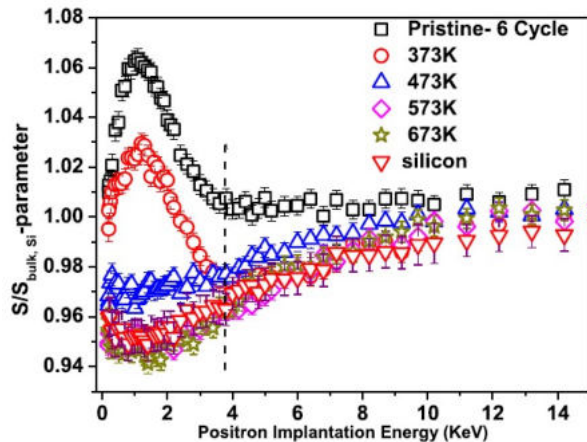


Element	At. %	Atomic Ratio	
		Experimental	Theoretical
Zn2p	5.69	1	1
N1s	17.62	3.1	4
O1s	7.06	1.2	0
C1s	69.62	10.2	8

- ZIF-8 nanocrystals surface (~ 3-4 nm) are enriched by Zn as compared to Imidazolate.
- Nanocrystals have different gas adsorption characteristics as compared to large size.



- Initial cycles: Annihilation from the particles surface or particle-substrate interface.
- Later cycles: Peak in S-parameter profiles shows positronium diffusion to interface
- Positronium diffusion length: ~ 1.4 micrometer which is consistent with literature.



On annealing: Collapsing of pore interconnectivity starts before the lattice decomposition

[Sharma et al. Mic. Meso. Mater. 307 \(2020\) 110519](#)



Conclusions and outlook:

- Positron annihilation spectroscopy is highly useful for experimental investigation of pore architecture
- ZIF-8 porosity depends on the external stimuli such as temperature, pressure and crystal size
- Pore size and porosity can be fine controlled using different strategies making ZIF-8 suitable for gas storage and gas separations.
- A systematic comparison of PAS data with other technique is required to establish the findings (gas adsorption does not work).
- ZIFs are proposed to be used in the form of membranes; pore aperture size using positron beam close to surface will be highly relevant for the gas separation efficiency.

Acknowledgement

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Dr. J. Bahadur

Mr. U. K. Goutam

Dr. P. K. Pujari

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